



(12) **EUROPEAN PATENT APPLICATION**

(43) Date of publication: 25.09.1996 Bulletin 1996/39  
 (51) Int. Cl.<sup>6</sup>: F01N 3/08, F01N 9/00  
 (21) Application number: 96104625.7  
 (22) Date of filing: 22.03.1996

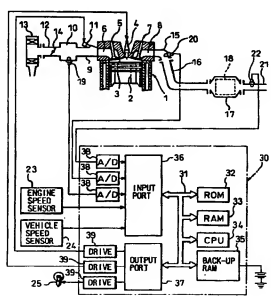
(84) Designated Contracting States:  
 DE FR GB  
 (30) Priority: 24.03.1995 JP 65947/95  
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(54) **An exhaust purification device of an engine**

(57) An exhaust purification device of an engine providing a NO<sub>x</sub> absorbent arranged in the exhaust passage. An O<sub>2</sub> sensor generating a current proportional to the air-fuel ratio is arranged in the engine exhaust passage downstream of the NO<sub>x</sub> absorbent. The amount of NO<sub>x</sub> actually absorbed in the NO<sub>x</sub> absorbent at the time of release of the NO<sub>x</sub> is calculated on the basis of the output signal of this O<sub>2</sub> sensor. On the basis of this calculated amount of NO<sub>x</sub>, correction is made so that the estimated amount of NO<sub>x</sub> represents the actual amount of absorption of NO<sub>x</sub>. When this corrected estimated amount of NO<sub>x</sub> reaches a set value, the action of releasing the NO<sub>x</sub> from the NO<sub>x</sub> absorbent is carried out.

Fig. 1



## Description

## BACKGROUND OF THE INVENTION

## 1. Field of the Invention

The present invention relates to an exhaust purification device of an engine.

## 2. Description of the Related Art

An internal combustion engine has been developed which provides in the engine exhaust passage an  $\text{NO}_x$  absorbent which absorbs  $\text{NO}_x$  when the air-fuel ratio of the inflowing exhaust gas is lean and releases the absorbed  $\text{NO}_x$  when the air-fuel ratio of the inflowing exhaust gas becomes rich, which estimates the amount of  $\text{NO}_x$  absorbed in the  $\text{NO}_x$  absorbent from the engine operating state, and, when the amount of  $\text{NO}_x$  estimated to be absorbed in the  $\text{NO}_x$  absorbent exceeds a predetermined set value, changes the air-fuel ratio of the exhaust gas flowing into the  $\text{NO}_x$  absorbent from lean to rich to make the  $\text{NO}_x$  be released from the  $\text{NO}_x$  absorbent.

This estimated amount of absorption of  $\text{NO}_x$ , however, does not always coincide with the actual amount of absorption of  $\text{NO}_x$ , that is, sometimes will be smaller or larger than the actual amount of absorption of  $\text{NO}_x$ . Accordingly, where the air-fuel ratio of the exhaust gas flowing into the  $\text{NO}_x$  absorbent is changed from lean to rich when the estimated amount of absorption of  $\text{NO}_x$  exceeds the predetermined set value, if the estimated amount of absorption of  $\text{NO}_x$  is smaller than the actual amount of absorption of  $\text{NO}_x$ , the ability of absorption of the  $\text{NO}_x$  absorbent is saturated before the estimated amount of  $\text{NO}_x$  reaches the set value, so there arises a problem in that the  $\text{NO}_x$  is not absorbed into the  $\text{NO}_x$  absorbent but is released into the atmosphere. In contrast, if the estimated amount of absorption of  $\text{NO}_x$  is larger than the actual amount of absorption of  $\text{NO}_x$ , the air-fuel ratio is made rich when the amount of  $\text{NO}_x$  absorbed in the  $\text{NO}_x$  absorbent is still small, and therefore the frequency of the air-fuel ratio being made rich becomes high, and thus there arises a problem that the amount of fuel consumption is increased.

## SUMMARY OF THE INVENTION

An object of the present invention is to provide an exhaust gas purification device capable of preventing  $\text{NO}_x$  from being released into the atmosphere and also preventing the fuel consumption from increasing.

According to the present invention, there is provided an exhaust purification device of an engine having an exhaust passage, provided with an  $\text{NO}_x$  absorbent arranged in the exhaust passage, the  $\text{NO}_x$  absorbent absorbing  $\text{NO}_x$  therein when an air-fuel ratio of exhaust gas flowing into the  $\text{NO}_x$  absorbent is lean and releasing absorbed  $\text{NO}_x$  therefrom when the air-fuel ratio of the

exhaust gas flowing into the  $\text{NO}_x$  absorbent becomes rich; an estimating means for estimating an amount of  $\text{NO}_x$  absorbed in the  $\text{NO}_x$  absorbent to obtain an estimated amount of  $\text{NO}_x$  stored in the  $\text{NO}_x$  absorbent; an air-fuel ratio detecting means arranged in the exhaust passage downstream of the  $\text{NO}_x$  absorbent for generating an output signal indicating an air-fuel ratio of exhaust gas which flows out from the  $\text{NO}_x$  absorbent; a  $\text{NO}_x$  amount calculating means for calculating an entire amount of  $\text{NO}_x$  stored in the  $\text{NO}_x$  absorbent on the basis of the output signal of the air-fuel ratio detecting means when the air-fuel ratio of exhaust gas flowing into the  $\text{NO}_x$  absorbent is changed from lean to rich so as to release  $\text{NO}_x$  from the  $\text{NO}_x$  absorbent; a correction value calculating means for calculating a correction value for the estimated amount of  $\text{NO}_x$ , which correction value is a value by which the estimated amount of  $\text{NO}_x$  corrected by the correction value when the air-fuel ratio of exhaust gas flowing into the  $\text{NO}_x$  absorbent is changed from lean to rich so as to release  $\text{NO}_x$  from the  $\text{NO}_x$  absorbent indicates the entire amount of  $\text{NO}_x$  calculated by the  $\text{NO}_x$  amount calculating means; and a control means for controlling the air-fuel ratio of exhaust gas flowing into the  $\text{NO}_x$  absorbent to change the air-fuel ratio of exhaust gas flowing into the  $\text{NO}_x$  absorbent from lean to rich to release  $\text{NO}_x$  from the  $\text{NO}_x$  absorbent when the estimated amount of  $\text{NO}_x$  corrected by the correction value exceeds a predetermined amount.

## BRIEF DESCRIPTION OF THE DRAWINGS

The present invention may be more fully understood from the description of the preferred embodiments of the invention set forth below with reference to the accompanying drawings, wherein:

Fig. 1 is an overall view of an engine;

Fig. 2 is a view of a map of a basic fuel injection time;

Fig. 3 is a view of a correction coefficient K;

Fig. 4 is a graph schematically showing a concentration of unburnt HC and CO and oxygen in exhaust gas discharged from the engine;

Figs. 5A and 5B are views for explaining an absorption and releasing action of  $\text{NO}_x$ ;

Fig. 6 is a view of an amount of absorption of  $\text{NO}_x$  NOXA;

Fig. 7 is a time chart of the air-fuel ratio control;

Figs. 8A and 8B are views of a cycle of making the air-fuel ratio of an air-fuel mixture rich for releasing  $\text{NO}_x$  and a rich time at this time;

Fig. 9 is a view of a current flowing between an anode and a cathode of an  $\text{O}_2$  sensor;

Figs. 10 and 11 are time charts showing the change of the value of a current flowing between the anode and cathode of the  $\text{NO}_x$  sensor;

Figs. 12 and 13 are flow charts of the control of the air-fuel ratio;

Fig. 14 is a flow chart of a feedback control I;

Fig. 15 is a time chart of the change of a feedback correction coefficient FAF;

Fig. 16 is a flow chart of a feedback control II;

Fig. 17 is a flow chart of processing for release of  $\text{NO}_x$ ;

Fig. 18 is a flow chart of a decision of deterioration;

Figs. 19A and 19B are views of a cycle TL for making the air-fuel ratio of the air-fuel mixture rich for releasing  $\text{NO}_x$  and the rich time TR;

Figs. 20 and 21 are flow charts of another embodiment for controlling the air-fuel ratio;

Fig. 22 is a flow chart for the processing for release of  $\text{NO}_x$ ; and

Fig. 23 is a flow chart of a decision of deterioration.

#### DESCRIPTION OF THE PREFERRED EMBODIMENTS

Referring to Fig. 1, 1 denotes an engine body, 2, a piston, 3, a combustion chamber, 4, a spark plug, 5, an intake valve, 6, an intake port, 7, an exhaust valve, and 8, an exhaust port. The intake port 6 is connected to a surge tank 10 via a corresponding branch pipe 9, and a fuel injector 11 injecting fuel toward the interior of the input port 6 is attached to each branch pipe 9. The surge tank 10 is connected via an intake duct 12 to an air cleaner 13, and a throttle valve 14 is arranged in the intake duct 12. On the other hand, the exhaust port 8 is connected via an exhaust manifold 15 and an exhaust pipe 16 to a casing 17 containing a  $\text{NO}_x$  absorbent 18.

An electronic control unit 30 comprises a digital computer and is provided with a read only memory (ROM) 32, a random access memory (RAM) 33, a microprocessor (CPU) 34, a back-up RAM 35 continuously connected to a power source, an input port 36, and an exhaust port 37 - all of which are connected to each other by a bi-directional bus 31. In the surge tank 10, a pressure sensor 19 for generating an output voltage proportional to an absolute pressure in the surge tank 10 is arranged. The output voltage of this pressure sensor 19 is input to the input port 36 via a corresponding analog-to-digital (AD) converter 38. An air-fuel ratio sensor (hereinafter referred to as an  $\text{O}_2$  sensor) 20 is arranged in the exhaust manifold 15, and the output of this  $\text{O}_2$  sensor 20 is input to the input port 36 via the corresponding AD converter 38. Another air-fuel ratio sensor (hereinafter referred to as an  $\text{O}_2$  sensor) 22 is arranged in the exhaust pipe 21 downstream of the  $\text{NO}_x$  absorbent 18. This  $\text{O}_2$  sensor 22 is connected to the input port 36 via a corresponding AD converter 38. Further, an engine speed sensor 23 generating an output pulse representing the engine speed and a vehicle speed sensor 24 generating an output pulse representing the vehicle speed are connected to the input port 36. On the other hand, the output port 37 is connected via the corresponding drive circuit 39 to the spark plug 4, fuel injection valve 11, and the alarm lamp 25.

In the engine shown in Fig. 1, a fuel injection time TAU is calculated on the basis of for example the following equation:

$$\text{TAU} = \text{TP} \cdot \text{K} \cdot \text{FAF}$$

Here, TP represents a basic fuel injection time, K, a correction coefficient, and FAF, a feedback correction coefficient, respectively. The basic fuel injection time TP indicates a fuel injection time necessary for making the air-fuel ratio of the air-fuel mixture to be supplied into the engine cylinder the stoichiometric air-fuel ratio. This basic fuel injection time TP is found in advance by experiments and preliminarily stored in the ROM 32 in the form of a map as shown in Fig. 2 as a function of the absolute pressure PM in the surge tank 10 and the engine rotation speed N. The correction coefficient K is a coefficient for controlling the air-fuel ratio of the air-fuel mixture to be supplied into the engine cylinder. If  $\text{K} = 1.0$ , the air-fuel ratio of the air-fuel mixture to be supplied into the engine cylinder becomes the stoichiometric air-fuel ratio. Contrary to this, when K becomes smaller than 1.0, the air-fuel ratio of the air-fuel mixture to be supplied into the engine cylinder becomes larger than the stoichiometric air-fuel ratio, that is, lean, and when K becomes larger than 1.0, the air-fuel ratio of the air-fuel mixture supplied into the engine cylinder becomes smaller than the stoichiometric air-fuel ratio, that is, rich.

The feedback correction coefficient FAF is a coefficient for making the air-fuel ratio accurately coincide with the stoichiometric air-fuel ratio on the basis of the output signal of the  $\text{O}_2$  sensor 20 when  $\text{K} = 1.0$ , that is, when the air-fuel ratio of the air-fuel mixture supplied into the engine cylinder should be made the stoichiometric air-fuel ratio. This feedback correction coefficient FAF moves up or down around about 1.0. The FAF is decreased when the air-fuel mixture becomes rich and increased when the air-fuel mixture becomes lean. Note that, when  $\text{K} < 1.0$  or  $\text{K} > 1.0$ , the FAF is fixed to 1.0.

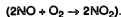
The target air-fuel ratio of the air-fuel mixture which should be supplied into the engine cylinder, that is, the value of the correction coefficient K, is changed in accordance with the operating state of the engine. In the embodiment according to the present invention, basically, as shown in Fig. 3, it is determined in advance as a function of the absolute pressure PM in the surge tank 10 and the engine speed N. Namely, as shown in Fig. 3, in a low load operation region on the lower load side from a solid line R, K becomes smaller than 1.0, that is, the air-fuel ratio of the air-fuel mixture is made lean, and in a high load operation region between the solid line R and solid line S, K becomes equal to 1.0, that is, the air-fuel ratio of the air-fuel mixture is made the stoichiometric air-fuel ratio. In the full load operation region on the higher load side from the solid line S, K becomes larger than 1.0, that is, the air-fuel ratio of the air-fuel mixture is made rich.

Figure 4 schematically shows the concentration of representative components in the exhaust gas discharged from the combustion chamber 3. As seen from Fig. 4, the concentration of the unburnt HC and CO in the exhaust gas discharged from the combustion chamber 3 is increased as the air-fuel ratio of the air-fuel mixture supplied into the combustion chamber 3 becomes rich, and the concentration of the oxygen  $O_2$  discharged from the combustion chamber 3 is increased as the air-fuel ratio of the air-fuel mixture supplied into the combustion chamber 3 becomes lean.

A  $NO_x$  absorbent 18 accommodated in the casing 17 uses for example alumina as the carrier. On this carrier, at least one element selected from alkali metals such as for example potassium K, sodium Na, lithium Li, and cesium Cs, alkali earth metals such as barium Ba or calcium Ca, and rare earth metals such as lanthanum La or yttrium Y and a precious metal such as platinum Pt are carried. When the ratio of the air and fuel (hydrocarbon) supplied into the engine intake passage and the exhaust passage upstream of the  $NO_x$  absorbent 18 is referred to as the air-fuel ratio of the inflowing exhaust gas into the  $NO_x$  absorbent 18, this  $NO_x$  absorbent 18 performs the action of absorbing and releasing  $NO_x$  so as to absorb the  $NO_x$  when the air-fuel ratio of the inflowing exhaust gas is lean and release the absorbed  $NO_x$  when the oxygen concentration in the inflowing exhaust gas is lowered. Note that, where the fuel (hydrocarbon) or the air is not supplied into the exhaust passage upstream of the  $NO_x$  absorbent 18, the air-fuel ratio of the flowing exhaust gas coincides with the air-fuel ratio of the air-fuel mixture supplied into the combustion chamber 3, and therefore, in this case, the  $NO_x$  absorbent 18 absorbs the  $NO_x$  when the air-fuel ratio of the air-fuel mixture supplied into the combustion chamber 3 is lean and releases the absorbed  $NO_x$  when the oxygen concentration in the air-fuel mixture supplied into the combustion chamber 3 is lowered.

When the  $NO_x$  absorbent 18 is arranged in the engine exhaust passage, this  $NO_x$  absorbent 18 actually performs the absorbing and releasing action of  $NO_x$ , but there are areas of uncertainty regarding the detailed mechanism of this absorbing and releasing action. However, it can be considered that this absorbing and releasing action is carried out by the mechanism as shown in Figs. 5A and 5B. Next, an explanation will be made of this mechanism by taking as an example a case where platinum Pt and barium Ba are carried on this carrier, but a similar mechanism is obtained even if an other precious metal or alkali metal, alkali earth metal, and rare earth metal are used.

Namely, when the inflowing exhaust gas becomes considerably lean, the oxygen concentration in the inflowing exhaust gas is greatly increased, and as shown in Fig. 5A, the oxygen  $O_2$  is deposited on the surface of the platinum Pt in the form of  $O_2$  or  $O^2$ . On the other hand, the NO in the inflowing exhaust gas reacts with  $O_2$  or  $O^2$  on the surface of the platinum Pt and becomes  $NO_2$



Subsequently, one part of the generated  $NO_2$  is absorbed into the absorbent while being oxidized on the platinum Pt and bonded to the barium oxide BaO while being diffused in the absorbent in the form of a nitric acid ion  $NO_3^-$  as shown in Fig. 5A. In this way,  $NO_x$  is absorbed into the  $NO_x$  absorbent 18.

So far as the oxygen concentration in the inflowing exhaust gas is high,  $NO_2$  is generated on the surface of the platinum Pt, and so far as the  $NO_x$  absorbing capability of the absorbent is not saturated, the nitric acid ion  $NO_3^-$  formed by absorption of  $NO_2$  into the absorbent is generated. Contrary to this, when the oxygen concentration in the flowing exhaust gas is lowered and the amount of generation of the  $NO_2$  is lowered, the reaction advances in a reverse direction



and thus the nitric acid ion  $NO_3^-$  in the absorbent is released from the absorbent in the form of  $NO_2$ . Namely, when the oxygen concentration in the flowing exhaust gas is lowered,  $NO_x$  will be released from the  $NO_x$  absorbent 18. As shown in Fig. 4, when the degree of leanness of the inflowing exhaust gas becomes low, the oxygen concentration in the inflowing exhaust gas is lowered, and therefore when the degree of the leanness of the inflowing exhaust gas is lowered, even if the air-fuel ratio of the inflowing exhaust gas is lean,  $NO_x$  will be released from the  $NO_x$  absorbent 18.

On the other hand, when the air-fuel ratio of the air-fuel mixture supplied into the combustion chamber 3 is made rich and the air-fuel ratio of the inflowing exhaust gas becomes rich, as shown in Fig. 4, a large amount of unburnt HC and CO are discharged from the engine. These unburnt HC and CO react with the oxygen  $O_2$  or  $O^2$  on the platinum Pt and are oxidized. Further, when the air-fuel ratio of the inflowing exhaust gas becomes rich, the oxygen concentration in the inflowing exhaust gas is extremely lowered, so  $NO_2$  is released from the absorbent. This  $NO_2$  reacts with the unburnt HC and CO and is reduced as shown in Fig. 5B. In this way, when  $NO_2$  no longer exists on the surface of the platinum Pt, the  $NO_2$  is successively released from the absorbent. Accordingly, when the air-fuel ratio of the inflowing exhaust gas is made rich, the  $NO_x$  will be released from the  $NO_x$  absorbent 18 in a short time.

Namely, when the air-fuel ratio of the inflowing exhaust gas is made rich, first of all, the unburnt HC and CO immediately react with  $O_2$  or  $O^2$  on the platinum Pt and are oxidized, and then even if the  $O_2$  or  $O^2$  on the platinum Pt is consumed, if the unburnt HC and CO still remain, the  $NO_x$  released from the absorbent and the  $NO_x$  discharged from the engine are reduced. Accordingly, if the air-fuel ratio of the inflowing exhaust gas is made rich, the  $NO_x$  absorbed in the  $NO_x$  absorbent 18 is released in a short time, and, in addition, this released  $NO_x$  is reduced, so it becomes possible to pre-

vent the  $\text{NO}_x$  from being discharged into the atmosphere.

As mentioned above, when the lean air-fuel mixture is burned,  $\text{NO}_x$  is absorbed into the  $\text{NO}_x$  absorbent 18. However, there is a limit to the  $\text{NO}_x$  absorbing ability of the  $\text{NO}_x$  absorbent 18. When the  $\text{NO}_x$  absorbing capability of the  $\text{NO}_x$  absorbent 18 is saturated, the  $\text{NO}_x$  absorbent 18 no longer can absorb the  $\text{NO}_x$ . Accordingly, it is necessary to release the  $\text{NO}_x$  from the  $\text{NO}_x$  absorbent 18 before the  $\text{NO}_x$  absorbing capability of the  $\text{NO}_x$  absorbent 18 is saturated. For this purpose, it is necessary to estimate to what degree the  $\text{NO}_x$  has been absorbed in the  $\text{NO}_x$  absorbent 18. Next, an explanation will be made of the estimation method of this amount of absorption of  $\text{NO}_x$ .

When the lean air-fuel mixture is burned, the higher the engine load, the larger the amount of  $\text{NO}_x$  discharged from the engine per unit time, so the amount of  $\text{NO}_x$  absorbed into the  $\text{NO}_x$  absorbent 18 per unit time is increased. Also, the higher the engine speed, the larger the amount of  $\text{NO}_x$  discharged from the engine per unit time, so the amount of  $\text{NO}_x$  absorbed into the  $\text{NO}_x$  absorbent 18 per unit time is increased. Accordingly, the amount of  $\text{NO}_x$  absorbed into the  $\text{NO}_x$  absorbent 18 per unit time becomes a function of the engine load and the engine speed. In this case, the engine load can be represented by the absolute pressure in the surge tank 10, so the amount of  $\text{NO}_x$  absorbed into the  $\text{NO}_x$  absorbent 18 per unit time becomes a function of the absolute pressure  $\text{PM}$  in the surge tank 10 and the engine speed  $N$ . Accordingly, in the embodiment according to the present invention, the amount of  $\text{NO}_x$  absorbed into the  $\text{NO}_x$  absorbent 18 per unit time is found in advance as a function of the absolute pressure  $\text{PM}$  and the engine speed  $N$  by experiments. These amounts of absorption of  $\text{NO}_x$ ,  $\text{NOXA}$  and  $\text{PM}$  are stored in advance in the ROM 32 in the form of a map shown in Fig. 6 as a function of  $\text{PM}$  and  $N$ .

On the other hand, as mentioned before, during the period where the  $\text{NO}_x$  is released from the  $\text{NO}_x$  absorbent 18, the unburnt  $\text{HC}$  and  $\text{CO}$  contained in the exhaust gas, that is, the excess fuel, is used for reducing the  $\text{NO}_x$  released from the  $\text{NO}_x$  absorbent 18, therefore the amount  $\text{NOXD}$  of  $\text{NO}_x$  released from the  $\text{NO}_x$  absorbent 18 per unit time becomes proportional to the amount of excess fuel supplied per unit time. Note that the amount  $Q_{\text{ex}}$  of excess fuel supplied per unit time can be represented by the following equation:

$$Q_{\text{ex}} = f_1 \cdot (K - 1.0) \cdot \text{TP} \cdot N$$

Here,  $f_1$  indicates a proportional constant,  $K$ , a correction coefficient,  $\text{TP}$  a basic fuel injection time, and  $N$ , an engine speed. On the other hand, when the proportional constant is  $f_2$ , the amount  $\text{NOXD}$  of  $\text{NO}_x$  released from the  $\text{NO}_x$  absorbent 18 per unit time can be represented by

$$\text{NOXD} = f_2 \cdot Q_{\text{ex}}$$

so if  $f = f_1 \cdot f_2$ , the amount  $\text{NOXD}$  of  $\text{NO}_x$  released from the  $\text{NO}_x$  absorbent 18 per unit time can be represented by the following equation:

$$\text{NOXD} = f \cdot (K - 1.0) \cdot \text{TP} \cdot N$$

As mentioned above, when a lean air-fuel mixture is burned, the amount of absorption of  $\text{NO}_x$  per unit time is represented by  $\text{NOXD}$ , and when a rich air-fuel mixture is burned, the amount of release of  $\text{NO}_x$  per unit time is represented by  $\text{NOXD}$ , therefore the amount  $\Sigma\text{NOX}$  of  $\text{NO}_x$  estimated to be absorbed in the  $\text{NO}_x$  absorbent 18 will be represented by the following equation:

$$\Sigma\text{NOX} = \Sigma\text{NOX} + \text{NOXA} - \text{NOXD}$$

Therefore, in the embodiment according to the present invention, as shown in Fig. 7, when the amount  $\Sigma\text{NOX}$  of the  $\text{NO}_x$  estimated to be absorbed in the  $\text{NO}_x$  absorbent 18, in practice, the corrected amount of estimation of  $\text{NO}_x$ ,  $\Sigma\text{NOXK}$  mentioned later, reaches the allowable maximum value  $\text{MAX}$ , the air-fuel ratio of the air-fuel mixture is temporarily made rich, whereby  $\text{NO}_x$  is released from the  $\text{NO}_x$  absorbent 18.

However,  $\text{SO}_x$  is contained in the exhaust gas, and not only  $\text{NO}_x$ , but also  $\text{SO}_x$  are absorbed into the  $\text{NO}_x$  absorbent 18. The absorbing mechanism of  $\text{SO}_x$  to the  $\text{NO}_x$  absorbent 18 can be considered to be the same as the absorption mechanism of  $\text{NO}_x$ .

Namely, similar to the explanation of the absorbing mechanism of  $\text{NO}_x$ , when the explanation is made by taking as an example a case where platinum  $\text{Pt}$  and barium  $\text{Ba}$  are carried on the carrier, as mentioned before, when the air-fuel ratio of the inflowing exhaust gas is lean, the oxygen  $\text{O}_2$  is deposited on the surface of the platinum  $\text{Pt}$  in the form of  $\text{O}_2^-$  or  $\text{O}^{2-}$ , and the  $\text{SO}_2$  in the inflowing exhaust gas reacts with the  $\text{O}_2^-$  or  $\text{O}^{2-}$  on the surface of the platinum  $\text{Pt}$  and becomes  $\text{SO}_3$ . Subsequently, one part of the generated  $\text{SO}_3$  is absorbed into the absorbent while being further oxidized on the platinum  $\text{Pt}$  and bonded to the barium oxide  $\text{BaO}$  while being diffused in the absorbent in the form of a sulfuric acid ion  $\text{SO}_4^{2-}$  and stable sulfate  $\text{BaSO}_4$  is generated.

However, this sulfate  $\text{BaSO}_4$  is stable and hard to decompose. Even if the air-fuel ratio of the air-fuel mixture is made rich for just a short time as shown in Fig. 7, most of the sulfate  $\text{BaSO}_4$  is not decomposed and remains as it is. Accordingly, the sulfate  $\text{BaSO}_4$  is increased in the  $\text{NO}_x$  absorbent 18 along with the elapse of time, and thus the maximum amount of absorption of  $\text{NO}_x$  which can be absorbed by the  $\text{NO}_x$  absorbent 18 will be gradually lowered along with the elapse of time. Namely, in other words, the  $\text{NO}_x$  absorbent 18 will gradually deteriorate along with the elapse of time. When the maximum amount of absorption of  $\text{NO}_x$  by the  $\text{NO}_x$  absorbent 18 is lowered, it is necessary to release the  $\text{NO}_x$  from the  $\text{NO}_x$  absorbent 18 in a period when the amount of absorption of the  $\text{NO}_x$  in the  $\text{NO}_x$  absorbent 18 is small. For this purpose, first, it becomes

necessary to correctly detect the maximum amount of absorption of  $\text{NO}_x$  possible by the  $\text{NO}_x$  absorbent 18, that is, the degree of deterioration of the  $\text{NO}_x$  absorbent 18.

In the embodiment according to the present invention, the maximum amount of absorption of  $\text{NO}_x$  possible by the  $\text{NO}_x$  absorbent 18, that is, the degree of deterioration of the  $\text{NO}_x$  absorbent 18, is detected from the air-fuel ratio detected by the  $\text{O}_2$  sensor 22. This will be explained later.

Namely, when the air-fuel ratio of the air-fuel mixture supplied into the combustion chamber 3 becomes rich, as shown in Fig. 4, the exhaust gas containing the oxygen  $\text{O}_2$  and the unburnt HC and CO is discharged from the combustion chamber 3, but this oxygen  $\text{O}_2$  and the unburnt HC and CO do not react much at all with each other, and thus this oxygen  $\text{O}_2$  passes through the  $\text{NO}_x$  absorbent 18 and is discharged from the  $\text{NO}_x$  absorbent 18. On the other hand, when the air-fuel ratio of the air-fuel mixture supplied into the combustion chamber 3 becomes rich,  $\text{NO}_x$  is released from the  $\text{NO}_x$  absorbent 18. At this time, the unburnt HC and CO contained in the exhaust gas is used for reducing the released  $\text{NO}_x$ , so during a period when the  $\text{NO}_x$  is released from the  $\text{NO}_x$  absorbent 18, no unburnt HC and CO are discharged from the  $\text{NO}_x$  absorbent 18. Accordingly, during a period when the  $\text{NO}_x$  is continuously released from the  $\text{NO}_x$  absorbent 18, the oxygen  $\text{O}_2$  is contained in the exhaust gas discharged from the  $\text{NO}_x$  absorbent 18, but no unburnt HC and CO are contained, therefore during this term, the air-fuel ratio of the exhaust gas discharged from the  $\text{NO}_x$  absorbent 18 becomes slightly lean.

Subsequently, when all of the  $\text{NO}_x$  absorbed in the  $\text{NO}_x$  absorbent 18 is released, the unburnt HC and CO contained in the exhaust gas are not used for the reduction of the  $\text{O}_2$  in the  $\text{NO}_x$  absorbent 18 but are discharged as they are from the  $\text{NO}_x$  absorbent 18. Accordingly, the air-fuel ratio of the exhaust gas discharged from the  $\text{NO}_x$  absorbent 18 becomes rich at this time. Namely, when all of the  $\text{NO}_x$  absorbed in the  $\text{NO}_x$  absorbent 18 is released, the air-fuel ratio of the exhaust gas discharged from the  $\text{NO}_x$  absorbent 18 changes from lean to rich. Accordingly, all of the  $\text{NO}_x$  absorbed in the  $\text{NO}_x$  absorbent 18 is released from the  $\text{NO}_x$  absorbent 18 during the time elapsing from when the air-fuel ratio of the exhaust gas flowing into the  $\text{NO}_x$  absorbent 18 is changed from lean to rich to when the air-fuel ratio of the exhaust gas discharged from the  $\text{NO}_x$  absorbent 18 becomes rich. Therefore, from this, the amount of  $\text{NO}_x$  absorbed in the  $\text{NO}_x$  absorbent 18 is seen. This will be explained in slightly detail more next.

The  $\text{O}_2$  sensor 22 shown in Fig. 1 comprises a cup-like cylindrical body made of zirconia arranged in the exhaust passage. An anode made of a thin platinum film is formed on an inside surface of this cylindrical body and a cathode made of a thin platinum film is formed on an outside surface of this cylindrical body, respectively. The cathode is covered by a porous layer. Constant volt-

age is applied between the cathode and anode. In this  $\text{O}_2$  sensor 22, as shown in Fig. 9, a current  $I$  (mA) proportional to the air-fuel ratio  $A/F$  flows between the cathode and anode. Note that, in Fig. 9,  $I_0$  indicates the current when the air-fuel ratio  $A/F$  is the stoichiometric air-fuel ratio ( $= 14.6$ ). As seen from Fig. 9, when the air-fuel ratio  $A/F$  is lean, the current  $I$  is increased as the air-fuel ratio  $A/F$  becomes larger within a range where  $I > I_0$ , and the current  $I$  becomes zero when the air-fuel ratio  $A/F$  becomes rich of almost 13.0 or less.

Figure 10 shows the change of the air-fuel ratio  $(A/F)_{in}$  of the exhaust gas flowing into the  $\text{NO}_x$  absorbent 18, the change of the current  $I$  flowing between the cathode and anode of the  $\text{O}_2$  sensor 22, and the change of the air-fuel ratio  $(A/F)$  of the exhaust gas flowing out from the  $\text{NO}_x$  absorbent 18. As shown in Fig. 10, when the air-fuel ratio  $(A/F)$  of the exhaust gas flowing into the  $\text{NO}_x$  absorbent 18 is changed from lean to rich and the  $\text{NO}_x$  releasing action from the  $\text{NO}_x$  absorbent 18 is started, the air-fuel ratio  $(A/F)_{out}$  of the exhaust gas flowing out from the  $\text{NO}_x$  absorbent 18 abruptly becomes small to near the stoichiometric air-fuel ratio, and therefore the current  $I$  is abruptly decreased to near  $I_0$ . Subsequently, in a term when the  $\text{NO}_x$  releasing action from the  $\text{NO}_x$  absorbent 18 is carried out, the air-fuel ratio  $(A/F)_{out}$  of the exhaust gas flowing out from the  $\text{NO}_x$  absorbent 18 is held in a slightly lean state, and therefore the current  $I$  is held at a value slightly larger than the  $I_0$ . Subsequently, when all of the  $\text{NO}_x$  absorbed in the  $\text{NO}_x$  absorbent 18 is released, the air-fuel ratio  $(A/F)$  of the exhaust gas flowing out from the  $\text{NO}_x$  absorbent 18 abruptly becomes small and becomes rich, and therefore the current  $I$  abruptly falls to zero.

Figure 11 shows the change of the current  $I$  where the amount of  $\text{NO}_x$  contained in the  $\text{NO}_x$  absorbent 18 differs. Note that, in Fig. 11, the numerical values indicate the amount of  $\text{NO}_x$  absorbed in the  $\text{NO}_x$  absorbent 18. As shown in Fig. 11, when the amount of  $\text{NO}_x$  absorbed in the  $\text{NO}_x$  absorbent 18 is different, along with this, an elapsed time  $t$  from when the air-fuel ratio  $(A/F)_{in}$  of the exhaust gas flowing into the  $\text{NO}_x$  absorbent 18 is changed from lean to rich to when the current  $I$  becomes almost zero changes. The smaller the amount of  $\text{NO}_x$  absorbed in the  $\text{NO}_x$  absorbent 18, the shorter this elapsed time.  $\text{NO}_x$  is continuously released from the  $\text{NO}_x$  absorbent 18 for almost this elapsed time  $t$ . If the entire amount of  $\text{NO}_x$  released during this elapsed time  $t$  is found, the entire amount of  $\text{NO}_x$  absorbed in the  $\text{NO}_x$  absorbent 18 will be seen.

Note that, as mentioned before, the amount of release of  $\text{NO}_x$  released from the  $\text{NO}_x$  absorbent 18 is represented by the following equation:

$$\text{NOXD} = I_1 \cdot (K - 1.0) \cdot TP \cdot N$$

Accordingly, if the total sum of the amount of release of  $\text{NO}_x$   $\text{NOXD}$  during the elapsed time  $t$  is found, the entire amount of  $\text{NO}_x$  actually absorbed in the  $\text{NO}_x$  absorbent 18 can be detected.

By the say, to detect the maximum amount of absorption of  $\text{NO}_x$  possible by the  $\text{NO}_x$  absorbent 18, that is, the degree of deterioration of the  $\text{NO}_x$  absorbent 18, at detection, the amount of absorption  $\Sigma\text{NOX}$  of  $\text{NO}_x$  of the  $\text{NO}_x$  absorbent 18 must become the maximum amount of absorption of  $\text{NO}_x$ . Namely, when assuming that the  $\text{VNO}_x$  indicated by the broken line in Fig. 7 is the maximum amount of absorption of  $\text{NO}_x$  which is actually possible, when the amount of absorption of  $\text{NO}_x$   $\Sigma\text{NOX}$  of the  $\text{NO}_x$  absorbent 18 is smaller than this  $\text{VNO}_x$ , even if all of the  $\text{NO}_x$  is released from the  $\text{NO}_x$  absorbent 18, the maximum amount of absorption of  $\text{NO}_x$   $\text{VNO}_x$  cannot be found. This is because the entire amount of  $\text{NO}_x$  released at this time is smaller than the maximum amount of absorption of  $\text{NO}_x$ .

Contrary to this, when the  $\text{NO}_x$  is released from the  $\text{NO}_x$  absorbent 18 when the absorbing capability of the  $\text{NO}_x$  absorbent 18 is saturated, the entire amount of  $\text{NO}_x$  released at this time represents the maximum amount of absorption of  $\text{NO}_x$   $\text{VNO}_x$ . Therefore, in the embodiment according to the present invention, a decision level SAT which is slightly larger than the value near the maximum amount of absorption of  $\text{NO}_x$   $\text{VNO}_x$  at present is set, and as shown in Fig. 7. When the amount of absorption of  $\text{NO}_x$   $\Sigma\text{NOX}$  of the  $\text{NO}_x$  absorbent 18 reaches this decision level SAT, the entire  $\text{NO}_x$  is released from the  $\text{NO}_x$  absorbent 18, whereby the actual amount of absorption of  $\text{NO}_x$   $\text{VNO}_x$ , that is, the degree of deterioration of the  $\text{NO}_x$  absorbent 18 at this time, is found.

Note that, as shown in Fig. 7, the allowable maximum value MAX with respect to the amount of  $\text{NO}_x$   $\Sigma\text{NOX}$  is set to a value smaller than the maximum amount of absorption of  $\text{NO}_x$   $\text{VNO}_x$ , and when the  $\Sigma\text{NOX}$  reaches the allowable maximum value MAX, the decision of deterioration of the  $\text{NO}_x$  absorbent 18 is not carried out, and only the action of releasing  $\text{NO}_x$  from the  $\text{NO}_x$  absorbent 18 is carried out. The frequency of only the action of releasing the  $\text{NO}_x$  from the  $\text{NO}_x$  absorbent 18 being carried out is higher than the frequency of the decision of deterioration of the  $\text{NO}_x$  absorbent 18 being carried out, and therefore for a period after the decision of deterioration of the  $\text{NO}_x$  absorbent 18 is carried out until the next decision of deterioration of the  $\text{NO}_x$  absorbent 18 is carried out, a number of actions of releasing  $\text{NO}_x$  are carried out.

The amount of absorption of  $\text{NO}_x$   $\Sigma\text{NOX}$  of the  $\text{NO}_x$  absorbent 18 is, however, an estimated amount as mentioned before, and therefore this amount of absorption of  $\text{NO}_x$   $\Sigma\text{NOX}$  does not always represent the actual amount of absorption of  $\text{NO}_x$ . In this case, if for example the amount of absorption of  $\text{NO}_x$   $\Sigma\text{NOX}$  indicates a considerably higher value than the actual amount of absorption of  $\text{NO}_x$ , even if the amount of absorption of  $\text{NO}_x$   $\Sigma\text{NOX}$  reaches the decision level SAT, the actual amount of absorption of  $\text{NO}_x$  does not reach the actual maximum amount of absorption of  $\text{NO}_x$   $\text{VNO}_x$ , and thus there arises a problem in that the actual maximum

amount of absorption of  $\text{NO}_x$   $\text{VNO}_x$  cannot be correctly detected.

Therefore, in the embodiment according to the present invention, a correction value KX with respect to the amount of absorption of  $\text{NO}_x$   $\Sigma\text{NOX}$  is introduced. Whenever the amount of absorption of  $\text{NO}_x$   $\Sigma\text{NOX}$  reaches the allowable maximum value MAX and the release of  $\text{NO}_x$  from the  $\text{NO}_x$  absorbent 18 is carried out, the actual amount of absorption of  $\text{NO}_x$   $\text{XNO}_x$  is calculated on the basis of the output signal of the  $\text{NO}_x$  sensor 22, and the correction value KX is updated on the basis of the following equation:

$$\text{KX} = \text{KX} \cdot (\text{XNO}_x / \Sigma\text{NOX})$$

In this case, the corrected estimated amount of  $\text{NO}_x$  is represented by  $\Sigma\text{NIXX}$  ( $= \text{KX} \cdot \Sigma\text{NOX}$ ). Namely, where for example the estimated amount of absorption of  $\text{NO}_x$   $\Sigma\text{NOX}$  becomes smaller than the actual amount of absorption of  $\text{NO}_x$   $\text{XNO}_x$ , the value of the correction value KX is increased with respect to the value of the correction value KX which has been used heretofore so that  $\Sigma\text{NIXX}$  ( $= \text{KX} \cdot \Sigma\text{NOX}$ ) coincides with  $\text{XNO}_x$ . Accordingly, in the embodiment according to the present invention, in actuality, not when the estimated amount of  $\text{NO}_x$   $\Sigma\text{NOX}$  reaches MAX, but when the corrected estimated amount of  $\text{NO}_x$   $\Sigma\text{NOX}$  reaches the allowable maximum value MAX, the action of releasing  $\text{NO}_x$  is carried out.

When the maximum amount of absorption of  $\text{NO}_x$   $\text{VNO}_x$  becomes small, that is, when the degree of deterioration of the  $\text{NO}_x$  absorbent 18 becomes high, the allowable maximum value MAX becomes small, and thus as seen from Fig. 7, a cycle at which the air-fuel ratio is made rich for releasing  $\text{NO}_x$  becomes short. Further, when the degree of deterioration of the  $\text{NO}_x$  absorbent 18 becomes high and the allowable maximum value MAX becomes small, the time required for the release of  $\text{NO}_x$  becomes short, so the time during which the air-fuel ratio is maintained rich becomes short. Accordingly, when the degree of deterioration of the  $\text{NO}_x$  absorbent 18 is low, as shown in Fig. 8A, a cycle  $t_1$  when the air-fuel ratio is made rich and the time  $t_2$  during which the air-fuel ratio is maintained rich are relatively long, and when the degree of deterioration of the  $\text{NO}_x$  absorbent 18 becomes high, as shown in Fig. 8B, a cycle when the air-fuel ratio is made rich and the time during which the air-fuel ratio is maintained rich become short.

As mentioned above, in the embodiment according to the present invention, the actual amount of  $\text{NO}_x$   $\text{VNO}_x$  and  $\text{XNO}_x$  are calculated on the basis of the current I flowing between the cathode and anode of the  $\text{O}_2$  sensor 22 and the air-fuel ratio is controlled for releasing  $\text{NO}_x$  on the basis of these values  $\text{VNO}_x$  and  $\text{XNO}_x$ . In this case, the current I flowing between the cathode and anode of the  $\text{O}_2$  sensor 22 is converted to a voltage and input into the input port 36. In the electronic control unit 30, this voltage is converted to the corresponding

current I again and the air-fuel ratio is controlled on the basis of the current value I.

Figure 12 and Fig. 13 show a routine for control of the air-fuel ratio. This routine is executed by interruption at every predetermined time interval.

Referring to Fig. 12 and Fig. 13, first of all, at step 100, a basic fuel injection time TP is calculated from the relationship shown in Fig. 2. Subsequently, at step 101, it is determined whether or not a decision of deterioration flag indicating that the degree of deterioration of the  $\text{NO}_x$  absorbent 18 should be decided has been set. When the decision of deterioration flag has not been set, the processing routine proceeds to step 102, where it is determined whether or not the  $\text{NO}_x$  releasing flag indicating that the  $\text{NO}_x$  should be released from the  $\text{NO}_x$  absorbent 18 has been set. When the  $\text{NO}_x$  releasing flag has not been set, the processing routine proceeds to step 103.

At step 103, the correction coefficient K is calculated on the basis of Fig. 3. Subsequently, at step 104, it is determined whether or not the correction coefficient K is 1.0. When  $K = 1.0$ , that is, when the air-fuel ratio of the air-fuel mixture should be made the stoichiometric air-fuel ratio, the processing routine proceeds to step 126, at which the feedback control I of the air-fuel ratio is carried out. This feedback control I is shown in Fig. 14. On the other hand, when K does not equal 1.0, the processing routine proceeds to step 105, at which it is determined whether or not the correction coefficient K is smaller than 1.0. When  $K < 1.0$ , that is, when the air-fuel ratio of the lean air-fuel mixture should be made lean, the processing routine proceeds to step 127, at which the feedback control II of the air-fuel ratio is carried out. This feedback control II is shown in Fig. 16. On the other hand, when K is not smaller than 1.0, the processing routine proceeds to step 106, at which FAF is fixed to 1.0, and then the processing routine proceeds to step 107. At step 107, the fuel injection time TAU is calculated on the basis of the following equation:

$$\text{TAU} = \text{TP} \cdot \text{K} \cdot \text{FAF}$$

Subsequently, at step 108, it is determined whether or not the correction coefficient K is smaller than 1.0. When  $K < 1.0$ , that is, when a lean air-fuel mixture should be burned, the processing routine proceeds to step 109, at which the amount of absorption of  $\text{NO}_x$  NOXA is calculated from Fig. 6. Subsequently, at step 110, the amount of absorption of  $\text{NO}_x$  NOXD is made zero, and then the processing routine proceeds to step 113. Contrary to this, at step 108, when  $K \geq 1.0$  is determined, that is, when an air-fuel mixture of the stoichiometric air-fuel ratio or the rich air-fuel mixture should be burned, the processing routine proceeds to step 111, at which the amount of absorption of  $\text{NO}_x$  NOXD is calculated on the basis of the following equation:

$$\text{NOXD} = f \cdot (\text{K} - 1) \cdot \text{TP} \cdot \text{N}$$

Subsequently, at step 112, the amount of absorption of  $\text{NO}_x$  NOXA is made zero, and then the processing routine proceeds to step 113. At step 113, the amount  $\Sigma\text{NOX}$  estimated to be absorbed in the  $\text{NO}_x$  absorbent 18 is calculated on the basis of the following equation.

$$\Sigma\text{NOX} = \Sigma\text{NOX} + \text{NOXA} - \text{NOXD}$$

Subsequently, at step 114, by multiplying the estimated amount of  $\text{NO}_x$   $\Sigma\text{NOX}$  by KX, the corrected estimated amount of  $\text{NO}_x$ , that is, the actual amount of  $\text{NO}_x$   $\Sigma\text{NOX}$  is calculated. Subsequently, at step 115, it is determined whether or not the  $\Sigma\text{NOX}$  becomes negative. When  $\Sigma\text{NOX}$  becomes smaller than 0, the processing routine proceeds to step 116, at which the  $\Sigma\text{NOX}$  is made zero. Subsequently, at step 117, a current vehicle speed SP is added to  $\Sigma\text{SP}$ . This  $\Sigma\text{SP}$  indicates the cumulative traveling distance of the vehicle. Subsequently, at step 118, it is determined whether or not the cumulative travelling distance  $\Sigma\text{SP}$  is larger than the set value  $\text{SP}_0$ . When  $\Sigma\text{SP} \leq \text{SP}_0$ , the processing routine proceeds to step 119, at which it is determined whether or not the  $\Sigma\text{NOX}$  exceeds the allowable maximum value MAX (Fig. 7). When  $\Sigma\text{NOX}$  becomes larger than MAX, the processing routine proceeds to step 120, at which the  $\text{NO}_x$  releasing flag is set.

On the other hand, when it is determined at step 118 that  $\Sigma\text{SP} > \text{SP}_0$ , the processing routine proceeds to step 121, at which it is determined whether or not the amount of  $\text{NO}_x$   $\Sigma\text{NOX}$  becomes larger than SAT (Fig. 7). When  $\Sigma\text{NOX}$  becomes larger than SAT, the processing routine proceeds to step 122, at which the decision of deterioration flag is set, and then at step 123,  $\Sigma\text{SP}$  is made zero.

When the decision of deterioration flag is set, the processing routine goes from step 101 to step 124, at which the decision of deterioration is carried out. This decision of deterioration is shown in Fig. 18. On the other hand, when the  $\text{NO}_x$  releasing flag is set, the processing routine proceeds from step 102 to step 125, at which the processing for release of  $\text{NO}_x$  is performed. This processing for release of  $\text{NO}_x$  is shown in Fig. 17.

Next, an explanation will be made of the feedback control I to be carried out at step 126 of Fig. 12, that is, the feedback control for maintaining the air-fuel ratio at the stoichiometric air-fuel ratio on the basis of the output signal of the  $\text{O}_2$  sensor 22 referring to Fig. 14 and Fig. 15.

As shown in Fig. 15, the  $\text{O}_2$  sensor 20 generates an output voltage V of about 0.9V when the air-fuel ratio of the air-fuel mixture is rich and generates an output voltage V of about 0.1V when the air-fuel ratio of the air-fuel mixture is lean. The feedback control I shown in Fig. 14 is carried out on the basis of the output signal of this  $\text{O}_2$  sensor 20.

Referring to Fig. 14, first of all, it is determined at step 130 whether or not the output voltage V of the  $\text{O}_2$  sensor 20 is smaller than a reference voltage Vr of



about 0.45V. When  $V \leq V_r$ , that is, when the air-fuel ratio is lean, the processing routine proceeds to step 131, at which the delay count CDL is decremented exactly by one. Subsequently, at step 132, it is determined whether or not the delay count CDL becomes smaller than the minimum value TDR. When CDL becomes smaller than TDR, the processing routine proceeds to step 133, at which the CLD is made TDR and then the processing routine proceeds to step 137. Accordingly, as shown in Fig. 15, when  $V$  becomes equal to or smaller than  $V_r$ , the delay count value CDL is gradually decreased, and subsequently, the CDL is maintained at the minimum value TDR.

On the other hand, when it is determined at step 130 that  $V > V_r$ , that is, when the air-fuel ratio is rich, the processing routine proceeds to step 134, at which the delay count CDL is incremented exactly by one. Subsequently, at step 135, it is determined whether or not the delay count CDL becomes larger than the maximum value TDL. When CDL becomes larger than TDL, the processing routine proceeds to step 136, at which the CDL is made TDL and then the processing routine proceeds to step 137. Accordingly, as shown in Fig. 15, when  $V$  becomes larger than  $V_r$ , the delay count CDL is gradually increased, and then CDL is maintained at the maximum value TDL.

At step 137, it is determined whether or not the sign of the delay count CDL is inverted from positive to negative or from negative to positive in a period from the previous processing cycle to this processing cycle. When the sign of the delay count CDL is inverted, the processing routine proceeds to step 138, at which it is determined whether or not it is an inversion from positive to negative, that is, whether or not it is an inversion from rich to lean. When it is an inversion from rich to lean, the processing routine proceeds to step 139, at which the rich skip value RSR is added to the feedback correction coefficient FAF and thus, as shown in Fig. 15, the FAF is abruptly increased exactly by the rich skip value RSR. Contrary to this, at the time of an inversion from lean to rich, the processing routine proceeds to step 140, at which the lean skip value RSL is subtracted from the FAF, and thus as shown in Fig. 15, the FAF is abruptly decreased exactly by the lean skip value RSL.

On the other hand, when it is determined at step 137 that the sign of the delay count CDL is not inverted, the processing routine proceeds to step 141, at which it is determined whether or not the delay count CDL is negative. When  $CDL \leq 0$ , the processing routine proceeds to step 142, at which the rich integration value KIR ( $KIR < RSR$ ) is added to the feedback correction coefficient FAF, and thus as shown in Fig. 15, the FAF is gradually increased. On the other hand, when  $CDL > 0$ , the processing routine proceeds to step 143, at which rich integration value KIL ( $KIL < RSL$ ) is subtracted from FAF, and thus the FAF is gradually decreased as shown in Fig. 15. In this way, the air-fuel ratio is controlled to the stoichiometric air-fuel ratio.

Next, an explanation will be made of the feedback control for maintaining the air-fuel ratio to the target lean air-fuel ratio corresponding to the correction coefficient  $K$  on the basis of the feedback control  $I$  carried out at step 127 of Fig. 12, that is, the current  $I$  of the  $O_2$  sensor 22, referring to Fig. 16.

Referring to Fig. 16, first of all, at step 150, the target current value  $I_0$  corresponding to the target lean air-fuel ratio is calculated from the relationship shown in Fig. 9. Subsequently, at step 151, it is determined whether or not the current  $I$  of the  $O_2$  sensor 22 is larger than the target current  $I_0$ . When  $I > I_0$ , the processing routine proceeds to step 152, at which a constant value  $\Delta F$  is added to the feedback correction coefficient FAF, and when  $I \leq I_0$ , the processing routine proceeds to step 153, at which the constant value  $\Delta F$  is subtracted from the feedback correction coefficient FAF. In this way, the air-fuel ratio is maintained at the target lean air-fuel ratio.

Next, an explanation will be made of the control for release of  $NO_x$  carried out at step 125 of Fig. 12 referring to Fig. 17.

Referring to Fig. 17, first of all, at step 160, the correction coefficient  $K$  is made a constant value  $KK$  of for example about 1.3. Subsequently, at step 161, the fuel injection time  $TAU$  is calculated on the basis of the following equation:

$$TAU = TP \cdot K$$

Accordingly, when the processing for release of  $NO_x$  is started, the feedback control of the air-fuel ratio is stopped, and the air-fuel ratio of the air-fuel mixture is made rich. Subsequently, at step 162, the amount of release  $NOXD$  of the  $NO_x$  released from the  $NO_x$  absorbent 18 per unit time is calculated as follows:

$$NOXD = f \cdot (K - 1.0) \cdot TP \cdot N$$

Subsequently, at step 163, the amount of release  $XNO_x$  of  $NO_x$  actually released from the  $NO_x$  absorbent 18 is calculated on the basis of the following equation. Note that, in the following equation,  $\Delta t$  represents the interval of the time interruption.

$$XNO_x = XNO_x + NOXD \cdot \Delta t$$

Subsequently, at step 164, it is determined whether or not the current  $I$  of the  $O_2$  sensor 22 becomes lower than the predetermined constant value  $\alpha$  (Fig. 11). When  $I$  becomes smaller than  $\alpha$ , the processing routine proceeds to step 165, at which it is determined whether or not the absolute value  $|XNO_x - \Sigma NKOX|$  of the difference between the actual amount of release of  $NO_x$   $XNO_x$  and the corrected estimated amount of absorption of  $NO_x$   $\Sigma NKOX$  is larger than the constant value  $\beta$ . When  $|XNO_x - \Sigma NKOX| \leq \beta$ , the processing routine jumps to step 167. Contrary to this, when  $|XNO_x - \Sigma NKOX| > \beta$ , the processing routine proceeds

to step 166, at which the correction value  $KX$  is corrected on the basis of the following equation:

$$KX = KX \cdot XNO_x / \Sigma NKX$$

Subsequently, at step 167, the  $NO_x$  releasing flag is reset, and thus the air-fuel ratio of the air-fuel mixture is changed to the air-fuel ratio determined according to the operating state at that time, usually lean. Subsequently, at step 168,  $XNO_x$  and  $\Sigma NOX$  are made zero.

Next, an explanation will be made of the decision of deterioration carried out at step 124 of Fig. 12 referring to Fig. 18.

Referring to Fig. 18, first of all, at step 170, the correction coefficient  $K$  is made the constant value  $KK$  of for example about 1.3. Subsequently, at step 171, the fuel injection time  $TAU$  is calculated on the basis of the following equation:

$$TAU = TP \cdot K$$

Accordingly, when the decision of deterioration is started, the feedback control of the air-fuel ratio is stopped, and the air-fuel ratio of the air-fuel mixture is made rich. Subsequently, at step 172, the amount of release  $NOXD$  of  $NO_x$  released from the  $NO_x$  absorbent 18 is calculated based on the following equation:

$$NOXD = f \cdot (K - 1.0) \cdot TP \cdot N$$

Subsequently, at step 173, the amount of release  $VNO_x$  of  $NO_x$  actually released from the  $NO_x$  absorbent 18 is calculated on the basis of the following equation. Note that, in the following equation,  $\Delta t$  represents the interval of the time interruption.

$$VNO_x = VNO_x + NOXD \cdot \Delta t$$

Subsequently, at step 174, it is determined whether or not the current  $I$  of the  $O_2$  sensor 22 becomes lower than the predetermined constant value  $\alpha$  (Fig. 11). When  $I$  becomes smaller than  $\alpha$ , the processing routine proceeds to step 175, at which by multiplying the  $VNO_x$  by a constant value larger than 1.0, for example 1.1, the decision level  $SAT$  ( $= 1.1 \cdot VNO_x$ ) is calculated. In this way, the decision level  $SAT$  is set to a value larger than  $VNO_x$ , so this  $VNO_x$  represents the maximum amount of absorption of  $NO_x$  possible by the  $NO_x$  absorbent 18. Namely, if  $VNO_x$  represents an amount of absorption of  $NO_x$  smaller than the maximum amount of absorption of  $NO_x$ , the decision level  $SAT$  becomes large whenever the decision of deterioration is carried out, and thus finally the  $VNO_x$  represents the maximum amount of absorption of  $NO_x$ , that is, the degree of deterioration of the  $NO_x$  absorbent 18.

So as to find the decision level  $SAT$ , of course it is also possible to multiply another numerical value other than 1.1 with  $VNO_x$ , and the decision level  $SAT$  can be found by multiplying any number of 1.0 or more with

$VNO_x$ . Note, if the numerical value to be multiplied with  $VNO_x$  is made too large, the time from when the amount of absorption of  $NO_x$  of the  $NO_x$  absorbent 18 becomes the maximum amount of absorption of  $NO_x$  to when the action of release of  $NO_x$  is carried out becomes too long, so the amount of  $NO_x$  discharged to the atmosphere is increased. Accordingly, it is not preferred that the numerical value to be multiplied with  $VNO_x$  be set too large. This numerical value is preferably about 1.3 or less.

When the decision level  $SAT$  is calculated at step 175, the processing routine proceeds to step 176, at which by multiplying a positive numerical value of 1.0 or less, for example 0.8, with the  $VNO_x$ , the allowable maximum value  $MAX$  ( $= 0.8 \cdot VNO_x$ ) is calculated. Namely, the allowable maximum value  $MAX$  is also updated in accordance with the degree of deterioration of  $NO_x$  absorbent 18. Subsequently, at step 177, it is determined whether or not the maximum amount of absorption of  $NO_x$   $VNO_x$  reaches the predetermined minimum value  $MIN$ . When  $VNO_x$  becomes smaller than  $MIN$ , the processing routine proceeds to step 178, at which the alarm lamp 25 is turned on. Subsequently, at step 179, the decision of deterioration flag is reset. When the decision of deterioration flag is reset, the air-fuel ratio of the air-fuel mixture is changed to the air-fuel ratio in accordance with the operating state at that time, usually lean. Subsequently, at step 180,  $VNO_x$  and  $\Sigma NOX$  are made zero.

Figure 19 to Fig. 23 show another embodiment. Also in this embodiment, the decision of deterioration of the  $NO_x$  absorbent 18 is carried out when the corrected amount of absorption of  $NO_x$   $\Sigma NKX$  exceeds the decision level  $SAT$ , but control for release of  $NO_x$  from the decision of deterioration to when the next decision of deterioration is carried out can be carried out by a simpler method compared with the first embodiment. Namely, in this embodiment, as shown in Figs. 19A and 19B, a cycle  $TL$  at which the air-fuel ratio of the air-fuel mixture is made rich so as to release the  $NO_x$  from the  $NO_x$  absorbent 18 and the rich time  $TR$  of the air-fuel mixture at this time are determined in accordance with the maximum amount of absorption of  $NO_x$   $VNO_x$ , that is, the degree of deterioration of the  $NO_x$  absorbent 18. Namely, as shown in Fig. 19A, the lower the maximum amount of absorption of  $NO_x$   $VNO_x$ , in other words, the larger the degree of deterioration of the  $NO_x$  absorbent 18, the shorter the cycle  $TL$  at which the air-fuel ratio of the air-fuel mixture is made rich, and as shown in Fig. 19B, the lower the maximum amount of absorption of  $NO_x$   $VNO_x$ , in other words, the larger the degree of deterioration of the  $NO_x$  absorbent 18, the shorter the rich time  $TR$  of the air-fuel mixture. Note that, the relationships shown in Figs. 19A and 19B are preliminarily stored in the ROM 32.

Figure 20 and Fig. 21 show the routine for control of the air-fuel ratio for this second embodiment. This routine is executed by interruption at every predetermined time interval.

Referring to Fig. 20 and Fig. 21, first of all, at step 200, the basic fuel injection time TP is calculated from the relationship shown in Fig. 2. Subsequently, at step 201, it is determined whether or not the decision of deterioration flag indicating that the degree of deterioration of the NO<sub>x</sub> absorbent 18 should be decided has been set. When the decision of deterioration flag has not been set, the processing routine proceeds to step 202, at which it is determined whether or not the NO<sub>x</sub> releasing flag indicating that the NO<sub>x</sub> should be released from the NO<sub>x</sub> absorbent 18 has been set. When the NO<sub>x</sub> releasing flag has not been set, the processing routine proceeds to step 203.

At step 203, the correction coefficient K is calculated on the basis of Fig. 3. Subsequently, at step 204, it is determined whether or not the correction coefficient K is 1.0. When K = 1.0, that is, when the air-fuel ratio of the air-fuel mixture is made the stoichiometric air-fuel ratio, the processing routine proceeds to step 228, at which the feedback control I of the air-fuel ratio is carried out. This feedback control I is shown in Fig. 14. On the other hand, when K is not equal to 1.0, the processing routine proceeds to step 205, at which it is determined whether or not the correction coefficient K is smaller than 1.0. When K < 1.0, that is, when the air-fuel ratio of the lean air-fuel mixture should be made lean, the processing routine proceeds to step 229, at which the feedback control II of the air-fuel ratio is carried out. This feedback control II is shown in Fig. 16. On the other hand, when K is not smaller than 1.0, the processing routine proceeds to step 206, at which the FAF is fixed to 1.0, and then the processing routine proceeds to step 207. At step 207, the fuel injection time TAU is calculated on the basis of the following equation:

$$TAU = TP \cdot K \cdot FAF$$

Subsequently, at step 208, it is determined whether or not the correction coefficient K is smaller than 1.0. When K < 1.0, that is, when the lean air-fuel mixture should be burned, the processing routine proceeds to step 209, at which the amount of absorption of NO<sub>x</sub> NOXA is calculated from Fig. 6. Subsequently, at step 210, the amount of release of NO<sub>x</sub> NOXD is made zero. Subsequently, at step 211, the interval Δt of the time interruption is added to the count value TC. Accordingly, this count TC represents the elapsed time.

At step 211, when the estimated time TC is calculated, the processing routine proceeds to step 215, at which the amount ΣNOX of NO<sub>x</sub> estimated to be absorbed in the NO<sub>x</sub> absorbent 18 is calculated on the basis of the following equation:

$$\Sigma NOX = \Sigma NOX + NOXA - NOXD$$

On the other hand, when it is determined at step 208 that K ≥ 1.0, that is, when the air-fuel mixture of stoichiometric air-fuel ratio or the rich air-fuel mixture should be burned, the processing routine proceeds to

step 212, at which the amount of release of NO<sub>x</sub> NOXD is calculated on the basis of the following equation:

$$NOXD = f \cdot (K - 1.0) \cdot TP \cdot N$$

Subsequently, at step 213, the amount of absorption of NO<sub>x</sub> NOXA is made zero, and then at step 214, the elapsed time TC is made zero. Subsequently, the processing routine proceeds to step 215, at which the estimated amount of NO<sub>x</sub> ΣNOX is calculated.

Subsequently, at step 216, by multiplying the estimated amount of NO<sub>x</sub> ΣNOX by the correction value KX, the corrected estimated amount of NO<sub>x</sub>, that is, the actual amount of NO<sub>x</sub> ΣNKX is calculated. Subsequently, at step 217, it is determined whether or not the ΣNOX becomes negative. When ΣNOX becomes smaller than 0, the processing routine proceeds to step 218, at which ΣNOX is made zero. Subsequently, at step 219, the current vehicle speed SP is added to ΣSP. This ΣSP indicates the cumulative travelling distance of the vehicle. Then, at step 220, it is determined whether or not the cumulative travelling distance ΣSP is larger than the set value SP<sub>0</sub>. When ΣSP ≤ SP<sub>0</sub>, the processing routine proceeds to step 221, at which it is determined whether or not the elapsed time TC exceeds the cycle TL shown in Fig. 19A in accordance with the maximum amount of absorption of NO<sub>x</sub> VNO<sub>x</sub>. When TC becomes larger than TL, the processing routine proceeds to step 222, at which the NO<sub>x</sub> releasing flag is set.

On the other hand, when it is determined at step 220 that ΣSP > SP<sub>0</sub>, the processing routine proceeds to step 223, at which it is determined whether or not the ΣNKX becomes larger than the decision level SAT (Fig. 7). When ΣNKX becomes larger than SAT, the processing routine proceeds to step 224, at which the decision of deterioration flag is set, and then, at step 225, ΣSP is made zero.

When the decision of deterioration flag is set, the processing routine goes from step 201 to step 226, at which the decision of deterioration is carried out. This decision of deterioration is shown in Fig. 23. On the other hand, when the NO<sub>x</sub> releasing flag is set, the processing routine goes from step 202 to step 227, at which the processing for release of NO<sub>x</sub> is carried out. This processing for release of NO<sub>x</sub> is shown in Fig. 22.

Next, an explanation will be made of the control for releasing NO<sub>x</sub> carried out at step 227 of Fig. 20 referring to Fig. 22.

Referring to Fig. 22, first of all, at step 230, the correction coefficient K is made the constant value KK of for example about 1.3. Subsequently, at step 231, the fuel injection time TAU is calculated on the basis of the following equation:

$$TAU = TP \cdot K$$

Accordingly, when the processing for release of NO<sub>x</sub> is started, the feedback control of the air-fuel ratio is stopped, and the air-fuel ratio of the air-fuel mixture is

made rich. Subsequently, at step 232, the amount of release NOXD of NO<sub>x</sub> released from the NO<sub>x</sub> absorbent 18 per unit time is calculated on the basis of the following equation:

$$\text{NOXD} = f_1 \cdot (K \cdot 1.0) \cdot TP \cdot N$$

Subsequently, at step 233, the amount of release XNO<sub>x</sub> of NO<sub>x</sub> actually released from the NO<sub>x</sub> absorbent 18 is calculated on the basis of the following equation. Note that, in the following equation, Δt represents the interval of the time interruption.

$$\text{XNO}_x = \text{XNO}_x + \text{NOXD} \cdot \Delta t$$

Subsequently, at step 234, it is determined whether or not a rich time TR shown in Fig. 19B in accordance with the maximum amount of absorption of NO<sub>x</sub> VNO<sub>x</sub> elapses from when the processing for release of NO<sub>x</sub> is started. When the rich time TR is elapsed, the processing routine proceeds to step 235, at which it is determined whether or not absolute value |XNO<sub>x</sub> - ΣNKX| of the difference between the actual amount of release of NO<sub>x</sub> XNO<sub>x</sub> and the corrected estimated amount of absorption of NO<sub>x</sub> ΣNKX is larger than the constant value β. When |XNO<sub>x</sub> - ΣNKX| ≤ β, the processing routine jumps to step 237. Contrary to this, when |XNO<sub>x</sub> - ΣNKX| > β, the processing routine proceeds to step 236, at which the correction value KX is corrected based on the following equation:

$$\text{KX} = \text{KX} \cdot \text{XNO}_x / \Sigma \text{NKX}$$

Subsequently, at step 237, the NO<sub>x</sub> releasing flag is reset, and thus the air-fuel ratio of the air-fuel mixture is changed from rich to the air-fuel ratio determined according to the operating state at that time, usually lean. Subsequently, at step 238, TC, XNO<sub>x</sub> and ΣNOX are made zero.

Next, an explanation will be made of the decision of deterioration carried out at step 226 of Fig. 20 referring to Fig. 23.

Referring to Fig. 23, first of all, at step 240, the correction coefficient K is made the constant value KK of for example about 1.3. Subsequently, at step 241, the fuel injection time TAU is calculated on the basis of the following equation:

$$\text{TAU} = \text{TP} \cdot K$$

Accordingly, when the decision of deterioration is started, the feedback control of the air-fuel ratio is stopped, and the air-fuel ratio of the air-fuel mixture is made rich. Subsequently, at step 242, the amount of release NOXD of NO<sub>x</sub> released from the NO<sub>x</sub> absorbent 18 per unit time is calculated on the basis of the following equation:

$$\text{NOXD} = f \cdot (K \cdot 1.0) \cdot \text{TP} \cdot N$$

Subsequently, at step 243, the amount of release VNO<sub>x</sub> of NO<sub>x</sub> actually released from the NO<sub>x</sub> absorbent 18 is calculated on the basis of the following equation. Note that, in the following equation, Δt represents the interval of the time interruption.

$$\text{VNO}_x = \text{VNO}_x + \text{NOXD} \cdot \Delta t$$

Subsequently, at step 244, it is determined whether or not the current value I of the O<sub>2</sub> sensor 22 becomes lower than the predetermined constant value α (Fig. 11). When I becomes smaller than α, the processing routine proceeds to step 245, at which by multiplying a constant value larger than 1.0, for example, 1.1, with VNO<sub>x</sub>, the decision level SAT (= 1.1 · VNO<sub>x</sub>) is calculated. In this case, as mentioned before, VNO<sub>x</sub> represents the maximum amount of absorption of NO<sub>x</sub>, that is, the degree of deterioration of the NO<sub>x</sub> absorbent 18. Subsequently, at step 246, on the basis of the maximum amount of absorption of NO<sub>x</sub> VNO<sub>x</sub>, a cycle TL making the air-fuel ratio of the air-fuel mixture rich is calculated from the relationship shown in Fig. 19A, and then at step 247, the rich time TR of the air-fuel mixture is calculated from the relationship shown in Fig. 19B on the basis of the maximum amount of absorption of NO<sub>x</sub> VNO<sub>x</sub>.

Subsequently, at step 248, it is determined whether or not the maximum amount of absorption of NO<sub>x</sub> VNO<sub>x</sub> becomes lower than the predetermined minimum value MIN. When VNO<sub>x</sub> becomes smaller than MIN, the processing routine proceeds to step 249, at which the alarm lamp 25 is turned on. Subsequently, at step 250, the decision of deterioration flag is reset. When the decision of deterioration flag is reset, the air-fuel ratio of the air-fuel mixture is changed to the air-fuel ratio in accordance with the operating state at that time, usually lean. Subsequently, at step 251, the VNO<sub>x</sub> and ΣNOX are made zero.

As mentioned above, according to the present invention, when the amount of NO<sub>x</sub> actually absorbed in the NO<sub>x</sub> absorbent 18 becomes the predetermined set value, the action of releasing NO<sub>x</sub> from the NO<sub>x</sub> absorbent is carried out. Accordingly, it is possible to prevent the NO<sub>x</sub> from not being absorbed into the NO<sub>x</sub> absorbent and being released into the atmosphere or the amount of the fuel consumption being increased as in the conventional case.

Further, in the present invention, the amount of NO<sub>x</sub> actually absorbed in the NO<sub>x</sub> absorbent is detected, and the degree of deterioration of the NO<sub>x</sub> absorbent is decided on the basis of this, so the degree of deterioration of the NO<sub>x</sub> absorbent can be correctly decided.

While the invention has been described by reference to specific embodiments chosen for purposes of illustration, it should be apparent that numerous modifications could be made thereto by those skilled in the art without departing from the basic concept and scope of the invention.

An exhaust purification device of an engine providing a NO<sub>x</sub> absorbent arranged in the exhaust passage. An O<sub>2</sub> sensor generating a current proportional to the air-fuel ratio is arranged in the engine exhaust passage downstream of the NO<sub>x</sub> absorbent. The amount of NO<sub>x</sub> actually absorbed in the NO<sub>x</sub> absorbent at the time of release of the NO<sub>x</sub> is calculated on the basis of the output signal of this O<sub>2</sub> sensor. On the basis of this calculated amount of NO<sub>x</sub>, correction is made so that the estimated amount of NO<sub>x</sub> represents the actual amount of absorption of NO<sub>x</sub>. When this corrected estimated amount of NO<sub>x</sub> reaches a set value, the action of releasing the NO<sub>x</sub> from the NO<sub>x</sub> absorbent is carried out.

#### Claims

1. An exhaust purification device of an engine having an exhaust passage, comprising:

an NO<sub>x</sub> absorbent arranged in the exhaust passage, said NO<sub>x</sub> absorbent absorbing NO<sub>x</sub> therein when an air-fuel ratio of exhaust gas flowing into the NO<sub>x</sub> absorbent is lean and releasing absorbed NO<sub>x</sub> therefrom when the air-fuel ratio of the exhaust gas flowing into the NO<sub>x</sub> absorbent becomes rich;

estimating means for estimating an amount of NO<sub>x</sub> absorbed in the NO<sub>x</sub> absorbent to obtain an estimated amount of NO<sub>x</sub> stored in the NO<sub>x</sub> absorbent;

air-fuel ratio detecting means arranged in the exhaust passage downstream of the NO<sub>x</sub> absorbent for generating an output signal indicating an air-fuel ratio of exhaust gas which flows out from the NO<sub>x</sub> absorbent;

NO<sub>x</sub> amount calculating means for calculating an entire amount of NO<sub>x</sub> stored in the NO<sub>x</sub> absorbent on the basis of the output signal of said air-fuel ratio detecting means when the air-fuel ratio of exhaust gas flowing into the NO<sub>x</sub> absorbent is changed from lean to rich so as to release NO<sub>x</sub> from the NO<sub>x</sub> absorbent;

correction value calculating means for calculating a correction value for said estimated amount of NO<sub>x</sub>, which correction value is a value by which said estimated amount of NO<sub>x</sub> corrected by said correction value when the air-fuel ratio of exhaust gas flowing into the NO<sub>x</sub> absorbent is changed from lean to rich so as to release NO<sub>x</sub> from the NO<sub>x</sub> absorbent indicates said entire amount of NO<sub>x</sub> calculated by said NO<sub>x</sub> amount calculating means; and

control means for controlling the air-fuel ratio of exhaust gas flowing into the NO<sub>x</sub> absorbent to change the air-fuel ratio of exhaust gas flowing into the NO<sub>x</sub> absorbent from lean to rich to release NO<sub>x</sub> from the NO<sub>x</sub> absorbent when said estimated amount of NO<sub>x</sub> corrected by

said correction value exceeds a predetermined amount.

2. An exhaust purification device as set forth in claim 1, wherein the NO<sub>x</sub> absorbent contains at least one component selected from alkali metals consisting of potassium, sodium, lithium, and cesium, alkali earth metals consisting of barium and calcium, and rare earth metals consisting of lanthanum and yttrium and platinum.

3. An exhaust purification device as set forth in claim 1, wherein said estimation means increases an NO<sub>x</sub> storage amount in accordance with the amount of absorption of NO<sub>x</sub> determined according to the engine operating state when the air-fuel ratio of the exhaust gas flowing into the NO<sub>x</sub> absorbent is lean, and then decreases the NO<sub>x</sub> storage amount in accordance with the NO<sub>x</sub> releasing amount determined according to the engine operating state when the air-fuel ratio of the exhaust gas flowing into the NO<sub>x</sub> absorbent is rich, thereby to find said NO<sub>x</sub> estimated amount.

4. An exhaust purification device as set forth in claim 3, wherein said amount of absorption of NO<sub>x</sub> determined according to the engine operating state is a function of the engine speed and the engine load.

5. An exhaust purification device as set forth in claim 3, wherein said NO<sub>x</sub> releasing amount determined according to the engine operating state is proportional to the excess fuel amount.

6. An exhaust purification device as set forth in claim 1, wherein said air-fuel ratio detection means generates an output signal indicating that the air-fuel ratio is slightly lean during a period for which the NO<sub>x</sub> is released from the NO<sub>x</sub> absorbent after the air-fuel ratio of the exhaust gas flowing into the NO<sub>x</sub> absorbent is changed from lean to rich and generates an output signal indicating that the air-fuel ratio is rich when the NO<sub>x</sub> releasing action from the NO<sub>x</sub> absorbent is completed.

7. An exhaust purification device as set forth in claim 6, wherein said air-fuel ratio detection means comprises an air-fuel ratio detection sensor which increases its output current proportional to the increase of the air-fuel ratio.

8. An exhaust purification device as set forth in claim 6, wherein said NO<sub>x</sub> amount calculation means decreases the NO<sub>x</sub> storage amount in accordance with the NO<sub>x</sub> releasing amount determined according to the engine operating state during a period from when the air-fuel ratio of the exhaust gas flowing into the NO<sub>x</sub> absorbent is changed from lean to rich to when said air-fuel ratio detection means gen-

erates the output signal indicating that the air-fuel ratio is rich, and thereby said entire amount of  $\text{NO}_x$  stored in the  $\text{NO}_x$  absorbent is calculated.

9. An exhaust purification device as set forth in claim 8, wherein said  $\text{NO}_x$  releasing amount determined according to the engine operating state is proportional to the excess fuel amount.
10. An exhaust purification device as set forth in claim 1, wherein when defining the estimated amount of  $\text{NO}_x$  estimated by said estimation means as  $\Sigma\text{NOX}$  and defining the correction value calculated by said correction value calculation means as  $\text{KX}$ , the estimated amount of  $\text{NO}_x$   $\Sigma\text{NKX}$  corrected by said correction value is represented by the following equation:

$$\Sigma\text{NKX} = \text{KX} \cdot \Sigma\text{NOX}$$

11. An exhaust purification device as set forth in claim 10, wherein when defining said entire amount of  $\text{NO}_x$  calculated by said  $\text{NO}_x$  amount calculation means as  $\text{XNO}_x$ , said correction value  $\text{KX}$  is updated on the basis of the following equation:

$$\text{KX} = \text{KX} \cdot \text{XNO}_x / \Sigma\text{NKX}$$

12. An exhaust purification device as set forth in claim 11, wherein when the difference between the estimated amount of  $\text{NO}_x$   $\Sigma\text{NKX}$  corrected by said correction value and said entire amount of  $\text{NO}_x$   $\text{XNO}_x$  is larger than a predetermined value, said correction value  $\text{KX}$  is updated.
13. An exhaust purification device as set forth in claim 1, wherein the predetermined amount in said control means is smaller than the maximum amount of absorption of  $\text{NO}_x$  of the  $\text{NO}_x$  absorbent.
14. An exhaust purification device as set forth in claim 1, wherein the predetermined amount in said control means is larger than the maximum amount of absorption of  $\text{NO}_x$  of the  $\text{NO}_x$  absorbent, and deterioration decision means deciding the degree of deterioration of the  $\text{NO}_x$  absorbent on the basis of said entire amount of  $\text{NO}_x$  calculated by said  $\text{NO}_x$  amount calculation means is provided.
15. An exhaust purification device as set forth in claim 14, wherein said predetermined amount is made larger than said entire amount of  $\text{NO}_x$  by exactly a predetermined proportion.
16. An exhaust purification device as set forth in claim 15, wherein said proportion is larger than 1.0 and smaller than 1.3.

17. An exhaust purification device as set forth in claim 14, wherein when said entire amount of  $\text{NO}_x$  becomes smaller than the predetermined amount, said deterioration decision means decides that the  $\text{NO}_x$  absorbent is deteriorated.
18. An exhaust purification device as set forth in claim 1, wherein said  $\text{NO}_x$  amount calculation means comprises first  $\text{NO}_x$  amount calculation means for calculating said entire amount of  $\text{NO}_x$  for only releasing  $\text{NO}_x$  from the  $\text{NO}_x$  absorbent and second  $\text{NO}_x$  amount calculation means for calculating said entire amount of  $\text{NO}_x$  for releasing  $\text{NO}_x$  from the  $\text{NO}_x$  absorbent and detecting the degree of deterioration of the  $\text{NO}_x$  absorbent, when the entire amount of  $\text{NO}_x$  is calculated by the first  $\text{NO}_x$  amount calculation means, the predetermined amount in said control means is made smaller than the maximum amount of absorption of  $\text{NO}_x$  of the  $\text{NO}_x$  absorbent, and when the entire amount of  $\text{NO}_x$  is calculated by the second  $\text{NO}_x$  amount calculation means, the predetermined amount in said control means is made larger than the maximum amount of absorption of  $\text{NO}_x$  of the  $\text{NO}_x$  absorbent.
19. An exhaust purification device as set forth in claim 18, wherein the frequency of that the entire amount of  $\text{NO}_x$  is calculated by the second  $\text{NO}_x$  amount calculation means is lower than the frequency of that the entire amount of  $\text{NO}_x$  is calculated by the first amount of  $\text{NO}_x$  calculation means.
20. An exhaust purification device as set forth in claim 18, wherein the entire amount of  $\text{NO}_x$  calculated by said second  $\text{NO}_x$  amount calculation means represents the maximum amount of absorption of  $\text{NO}_x$   $\text{VNO}_x$  of the  $\text{NO}_x$  absorbent, when the entire amount of  $\text{NO}_x$  is calculated by the first  $\text{NO}_x$  amount calculation means, the predetermined amount is made smaller than the maximum amount of absorption of  $\text{NO}_x$   $\text{VNO}_x$  by exactly the predetermined proportion, and when the entire amount of  $\text{NO}_x$  is calculated by the second  $\text{NO}_x$  amount calculation means, the predetermined amount is made larger than the maximum amount of absorption of  $\text{NO}_x$   $\text{VNO}_x$  by exactly the predetermined proportion.
21. An exhaust purification device as set forth in claim 20, wherein deterioration decision means is provided for deciding the degree of deterioration of the  $\text{NO}_x$  absorbent on the basis of said maximum amount of absorption of  $\text{NO}_x$   $\text{VNO}_x$ .

Fig. 1

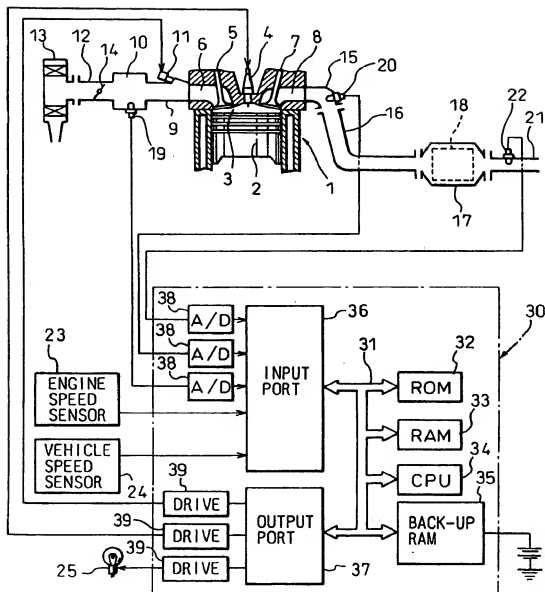


Fig. 2

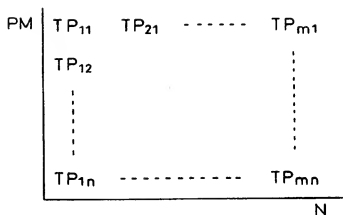


Fig. 3

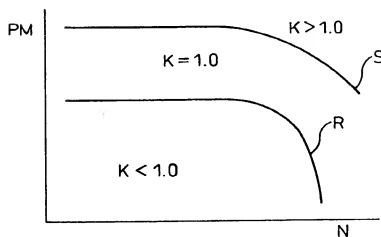




Fig. 4

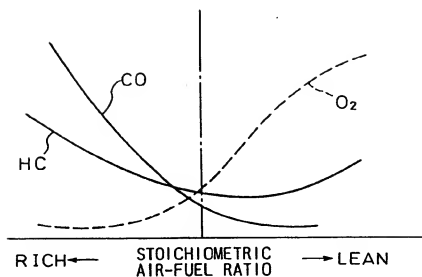


Fig. 5A

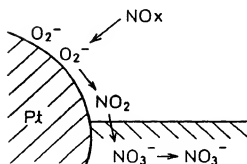


Fig. 5B

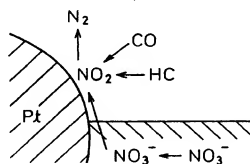


Fig.6

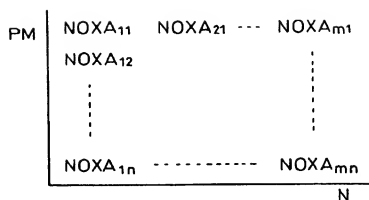
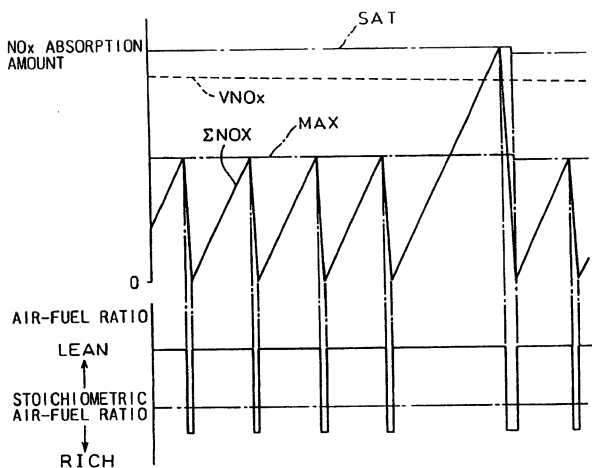


Fig. 7



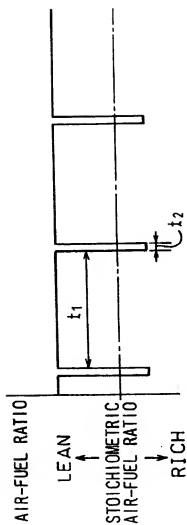


Fig. 8A

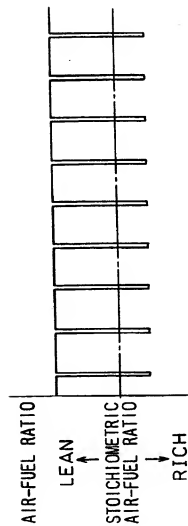


Fig. 8B

Fig. 9

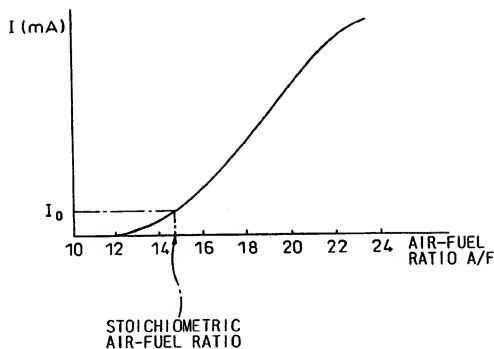


Fig.10

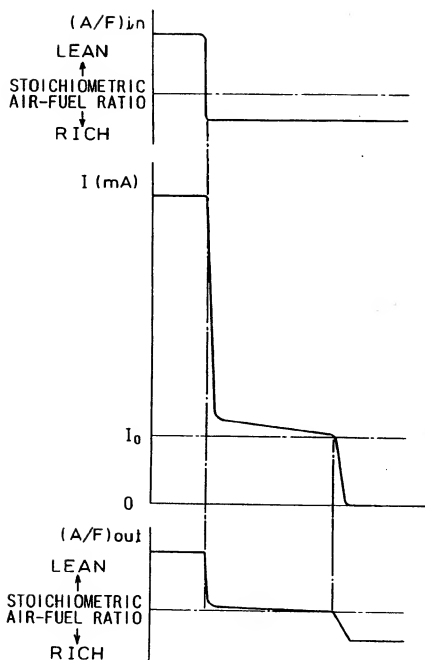


Fig.11

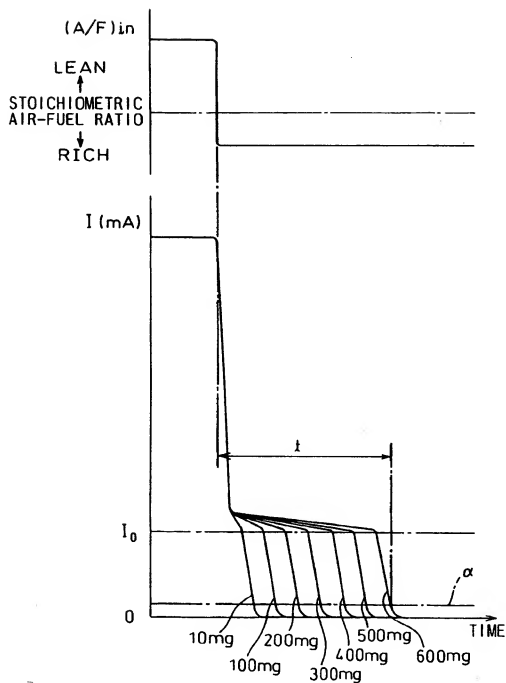


Fig.12

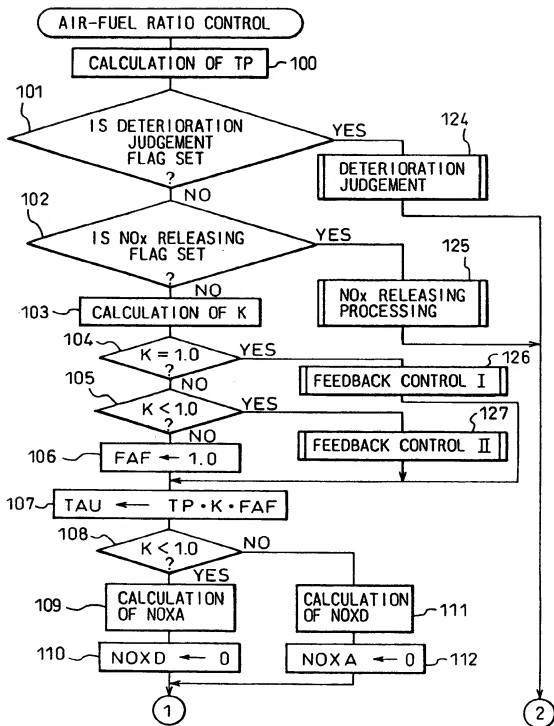




Fig.13

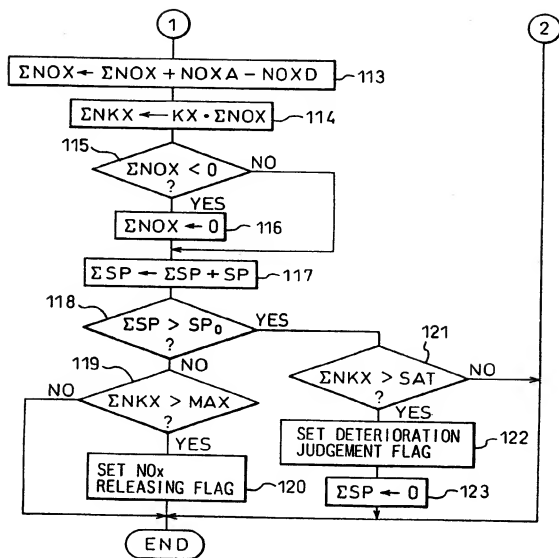


Fig.14

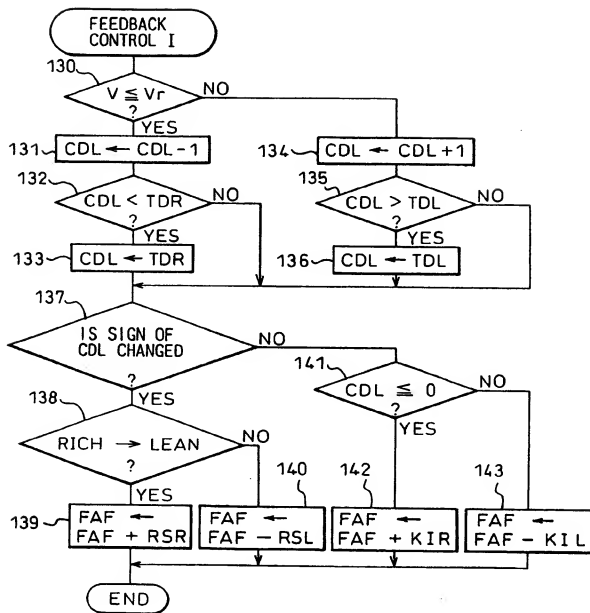


Fig.15

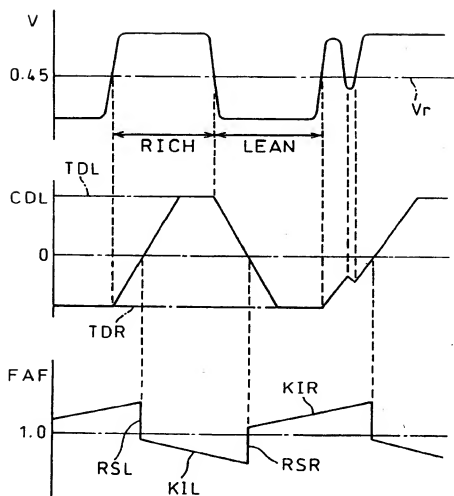


Fig.16

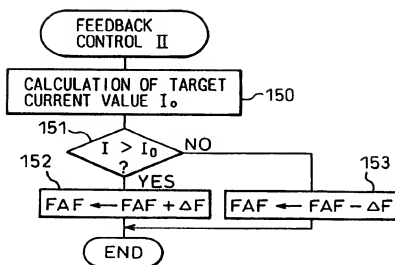


Fig.17

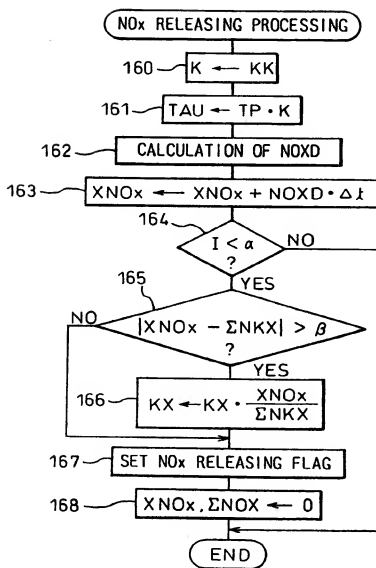


Fig.18

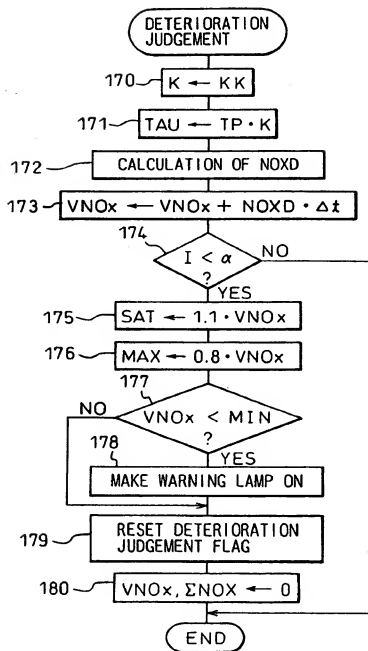


Fig.19A

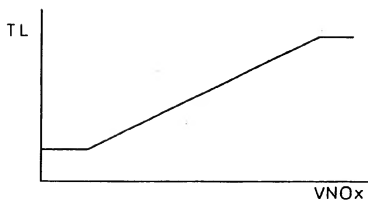


Fig.19B

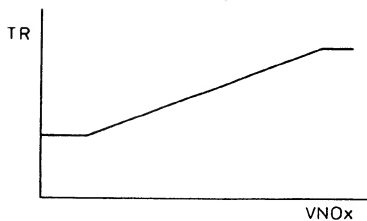


Fig. 20

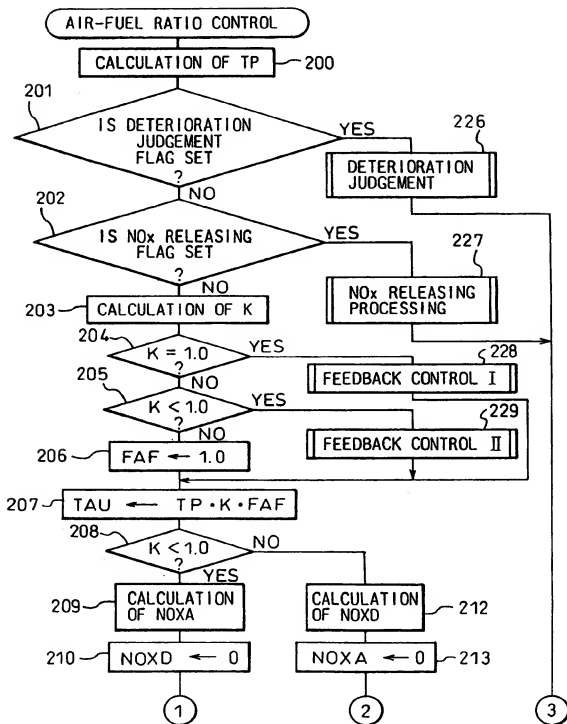




Fig.21

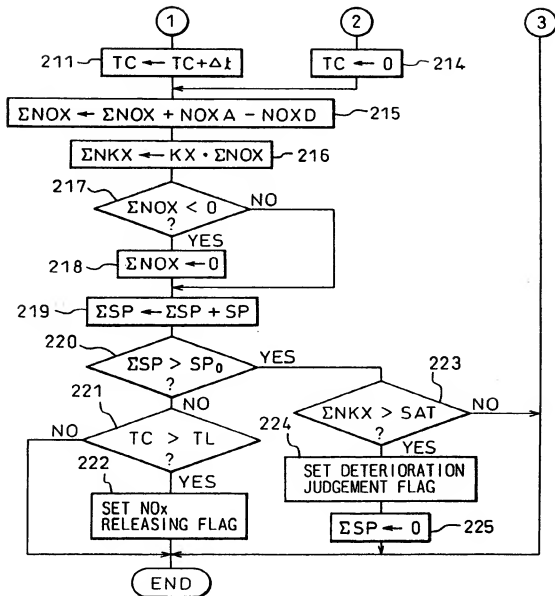


Fig. 22

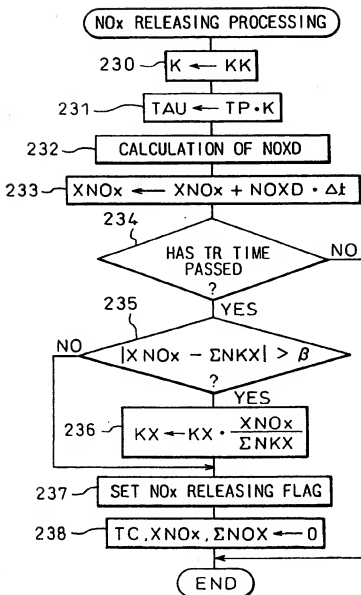
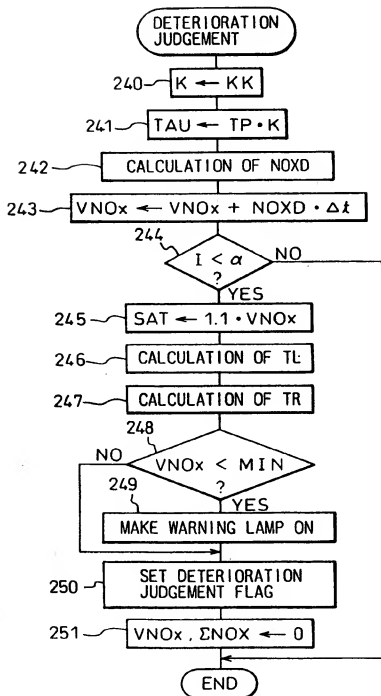


Fig. 23





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(11) EP 0 733 787 A3

(12)

# EUROPEAN PATENT APPLICATION

(88) Date of publication A3:  
28.05.1997 Bulletin 1997/22

(51) Int. Cl.<sup>6</sup>: F01N 3/08, F01N 9/00

(43) Date of publication A2:  
25.09.1996 Bulletin 1996/39

(21) Application number: 96104625.7

(22) Date of filing: 22.03.1996

(84) Designated Contracting States:  
DE FR GB

(30) Priority: 24.03.1995 JP 65947/95

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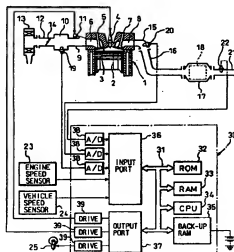
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(54) An exhaust purification device of an engine

(57) An exhaust purification device of an engine providing a NO<sub>x</sub> absorbent (18) arranged in the exhaust passage (16). An O<sub>2</sub> sensor (22) generating a current proportional to the air-fuel ratio is arranged in the engine exhaust passage downstream of the NO<sub>x</sub> absorbent. The amount of NO<sub>x</sub> actually absorbed in the NO<sub>x</sub> absorbent at the time of release of the NO<sub>x</sub> is calculated on the basis of the output signal of this O<sub>2</sub> sensor. On the basis of this calculated amount of NO<sub>x</sub>, correction is made so that the estimated amount of NO<sub>x</sub> represents the actual amount of absorption of NO<sub>x</sub>. When this corrected estimated amount of NO<sub>x</sub> reaches a set value, the action of releasing the NO<sub>x</sub> from the NO<sub>x</sub> absorbent is carried out.

Fig. 1





European Patent  
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## EUROPEAN SEARCH REPORT

Application Number  
EP 96 10 4625

DOCUMENTS CONSIDERED TO BE RELEVANT			
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int.Cl.6)
A	EP 0 636 770 A (TOYOTA MOTOR CO LTD) 1 February 1995 * column 2, line 34 - line 56 * * column 5, line 50 - column 6, line 20; figure 1 *	1,2	F01N3/08 F01N9/00
A	--- PATENT ABSTRACTS OF JAPAN vol. 018, no. 691 (M-1731), 26 December 1994 & JP 06 272546 A (TOYOTA MOTOR CORP), 27 September 1994, * abstract *	1,14	
A	--- PATENT ABSTRACTS OF JAPAN vol. 94, no. 010 & JP 06 280550 A (TOYOTA MOTOR CORP), 4 October 1994, * abstract *	1	
A	--- PATENT ABSTRACTS OF JAPAN vol. 018, no. 691 (M-1731), 26 December 1994 & JP 06 272540 A (TOYOTA MOTOR CORP), 27 September 1994, * abstract *	1	TECHNICAL FIELDS SEARCHED (Int.Cl.6) F01N
The present search report has been drawn up for all claims			
Place of search THE HAGUE		Date of completion of the search 25 March 1997	Examiner Sideris, M
CATEGORY OF CITED DOCUMENTS X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure P : intermediate document T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application L : document cited for other reasons & : member of the same patent family, corresponding document			

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